

## Lewis Acid Catalyzed, Selective Cyclopropane-Ring Opening in Ingol Diterpene Derivatives

by Masuna Srinivasulu<sup>a)</sup>, Vanimireddy Lakshmi Niranjan Reddy<sup>a)</sup>, Samala Malla Reddy<sup>a)</sup>, Veluri Ravikanth<sup>b)</sup>, Tuniki Venugopal Raju<sup>a)</sup>, Sistla Ramakrishna<sup>b)</sup>, and Yenamandra Venkateswarlu<sup>a)</sup>

<sup>a)</sup> Natural Products Laboratory, Organic Chemistry Division-I

<sup>b)</sup> Pharmacology Division, Indian Institute of Chemical Technology, Hyderabad-500007, India  
(phone: +91-40-27193167; fax: +91-40-27160512; e-mail: luchem@iict.res.in)

Lewis acid mediated skeletal rearrangement of the ingol diterpenoids **1** and **4** via regio- and stereospecific cyclopropane-ring opening afforded the four new compounds **2**, **3**, **5**, and **6**, named nivulianol A–D (*Scheme 1*). Their structures were established by means of IR, MS, and in-depth NMR spectroscopic analyses. The rearranged congeners were tested for lipopolysaccharide (LPS)-induced prostaglandin (PG) E<sub>2</sub> (cyclooxygenase-2) inhibition. Thereby, nivulianol B (= (1*S*\*,2*E*,4*R*\*,5*S*\*,7*Z*,9*S*\*,11*R*,13*S*\*,14*S*\*)-14-acetoxy-5-methoxy-3,9,13-trimethyl-6-(1-methylethenyl)-10-oxo-15-oxatricyclo[9.3.1.0<sup>11</sup>]pentadeca-2,7-dien-4-yl (2*Z*)-2-methylbut-2-enoate; **3**) was found to be significantly active, with an IC<sub>50</sub> value of 36.3 µg/ml.

**Introduction.** – The unique reactivity of cyclopropanes is due to their high level of ring strain, which offers considerable utility in organic synthesis [1–3]. Many potential applications of cyclopropanes as useful building blocks have been predicted based on regio- and stereo-controlled ring-opening reactions [4–6]. Typically, electrophilic ring opening has been achieved by transition metals (Pd, Pt, Rh, and Ir), halide ions, and Lewis acids [7–12]. Electron-donor-substituted cyclopropanes, particularly alkoxy, silyloxy, and arylsulfanyl compounds, have found increasing use, primarily due to facile and regio-controlled cyclopropane ring opening [13–18].

**Results and Discussion.** – 1. *Chemical Transformation and Structure Elucidation.* During our studies on the chemical modification of various secondary metabolites from plant sources [19] and marine organisms [20][21], we observed that BF<sub>3</sub>·OEt<sub>2</sub> reacts with the diterpenoid **1** (= 3,12-di-*O*-acetyl-8-*O*-methyl-7-*O*-(2-methylbut-2-enoyl)ingol)<sup>1)</sup> [22] to yield the rearranged compounds nivulianol A (**2**) and B (**3**), whereas scandium triflate, Sc(OTf)<sub>3</sub><sup>2)</sup>, gave only **3** (*Scheme 1*). In turn, the related compound **4** [23][24], lacking the angeloyl group<sup>3)</sup>, reacted with BF<sub>3</sub>·OEt<sub>2</sub> to afford both nivulianol C (**5**) and D (**6**), whereas, in the presence of Sc(OTf)<sub>3</sub>, only **6** was obtained.

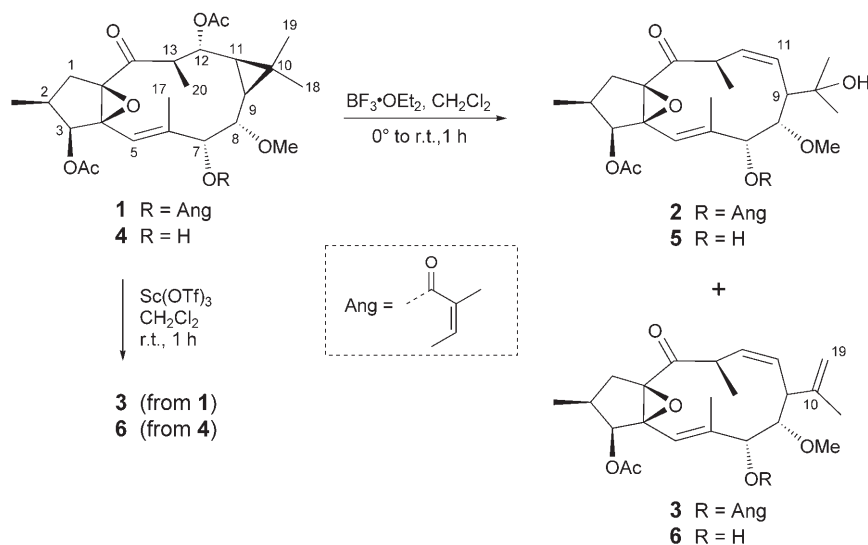
Compound **2** was obtained as a semisolid material ( $[\alpha]_{\text{D}} = -102.6$  ( $c = 0.15$ , CHCl<sub>3</sub>)), which analyzed for C<sub>28</sub>H<sub>40</sub>O<sub>8</sub> by HR-FAB-MS, as confirmed by the ion at  $m/z$  505.2765 ( $[M + 1]^+$ , calc. 505.2791). Its IR spectrum showed bands at 3423, 1740, 1720,

<sup>1)</sup> The systematic name of ingol is (1*R*,3*R*,4*R*,5*R*,7*S*,8*S*,9*R*,10*E*,12*S*,13*S*,14*S*)-4,8,9,13-tetrahydroxy-3,6,6,10,14-pentamethyl-16-oxatetracyclo[10.3.1.0<sup>1,12</sup>.0<sup>3,7</sup>]hexadec-10-en-2-one.

<sup>2)</sup> Triflate (Tf) = trifluoromethanesulfonate.

<sup>3)</sup> Angeloyl (Ang) = (*Z*)-2-methylbut-2-enoyl.

Scheme 1



and  $1705\text{ cm}^{-1}$ , indicating OH, and keto and ester  $\text{C}=\text{O}$  groups. The  $^1\text{H-NMR}$  spectrum of compound **2** (Table) showed an AcO group at  $\delta(\text{H})$  2.10 (*s*, 3 H), a MeO group at 3.48 (*s*, 3 H), and an AngO group ( $\delta(\text{H})$  6.15 (*m*, 1 H), 1.95 (*d*,  $J = 1.4$  Hz, 3 H), 2.03 (*dd*,  $J = 7.3, 1.4$  Hz, 3 H)). Also observed were three olefinic H-atoms at  $\delta(\text{H})$  5.81 (*s*, H–C(5)), 4.99 (*dd*,  $J = 14.9, 9.8$  Hz, H–C(12)), and 4.78 (*dd*,  $J = 14.9, 9.8$  Hz, H–C(11)), a vinylic Me group at 1.90 (*s*, Me(17)), two Me groups connected to an oxygenated C-atom at  $\delta(\text{H})$  1.15 and 0.99 (Me(18) and Me(19), resp.), two methine groups (geminal to ester functions) at  $\delta(\text{H})$  5.02 (*d*,  $J = 8.4$  Hz, H–C(3)), and 5.26 (*br. s*, H–C(7)), and a methine geminal to an ether linkage at  $\delta(\text{H})$  2.77 (*dd*,  $J = 12.0, 2.8$  Hz, H–C(8)). Finally, an exchangeable H-atom at  $\delta(\text{H})$  4.58 (*s*, 10-OH) was detected.

Careful comparison of the  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR (DEPT) spectra of compounds **1** and **2** (Table) revealed that the latter was devoid of signals due to the cyclopropane ring [ $\delta(\text{C})$  27.31 (C(9)), 19.28 (C(10)), 30.53 (C(11)) in **1**] and the methine connected to the Ac group ( $\delta(\text{C})$  71.0 (C(12) in **1**). Instead, the presence of an (*E*)-configured  $\text{C}=\text{C}$  bond at  $\delta(\text{C})$  132.31 (C(11)) and 129.8 (C(12)), and a quaternary C-atom connected to an O-atom at  $\delta(\text{C})$  81.4 (C(10)) were noticed. Further, the  $^1\text{H}, ^1\text{H}$ -COSY spectrum of **2** established that H–C(12) at  $\delta(\text{H})$  4.99 (*dd*) was correlated with H–C(13) at 3.68 (*dd*) and H–C(11) at 4.78 (*dd*). The latter, in turn, showed a correlation with H–C(9) at  $\delta(\text{H})$  3.69 (*dd*), which was linearly correlated with H–C(8) at  $\delta(\text{H})$  2.77 (*dd*).

The relative configurations at most of the stereogenic centers of **2** were established by NOESY experiments. The H–C(9) signal at  $\delta(\text{H})$  3.69 was correlated with Me(18) at  $\delta(\text{H})$  1.15 (*s*). H–C(7) at  $\delta(\text{H})$  5.26 (*s*) showed correlations with H–C(8), H–C(13), and Me(17) at  $\delta(\text{H})$  2.77 (*dd*), 3.68 (*dd*), and 1.90 (*s*), respectively. From all these data, the structure of nivulianol A (**2**) was established as (1*S*\*,2*E*,4*R*\*,5*S*\*,7*Z*,9*S*\*,11*R*\*,13*S*\*,14*S*\*)-14-acetoxy-6-(1-hydroxy-1-methylethyl)-

Table. NMR Data of Compounds **2**, **3**, **5**, and **6**. At 200 MHz in CDCl<sub>3</sub>;  $\delta$  in ppm,  $J$  in Hz.

Position	<b>2</b>		<b>3</b>		<b>5</b>		<b>6</b>	
	<sup>1</sup> H	<sup>13</sup> C	<sup>1</sup> H	<sup>13</sup> C	<sup>1</sup> H	<sup>13</sup> C	<sup>1</sup> H	<sup>13</sup> C
1 $\alpha$	1.74 ( <i>d</i> , $J = 14.6$ )	33.5	1.73 ( <i>d</i> , $J = 14.8$ )	33.82	1.73 ( <i>d</i> , $J = 14.5$ )	33.61	1.73 ( <i>d</i> , $J = 14.5$ )	33.86
1 $\beta$	2.74 ( <i>dd</i> , $J = 14.6, 9.1$ )	–	2.73 ( <i>dd</i> , $J = 14.8, 9.0$ )	–	2.80 ( <i>dd</i> , $J = 14.5, 9.0$ )	–	2.76 ( <i>dd</i> , $J = 14.5, 9.2$ )	–
2	2.50–2.62 ( <i>m</i> )	29.7	2.50–2.62 ( <i>m</i> )	29.79	2.50–2.62 ( <i>m</i> )	29.69	2.50–2.62 ( <i>m</i> )	29.69
3	5.02 ( <i>d</i> , $J = 8.4$ )	77.7	5.02 ( <i>d</i> , $J = 8.4$ )	79.34	5.10 ( <i>d</i> , $J = 8.4$ )	77.96	5.09 ( <i>d</i> , $J = 8.4$ )	78.10
4	–	73.7	–	73.72	–	73.73	–	73.72
5	5.81 ( <i>s</i> )	116.6	5.81 ( <i>s</i> )	116.78	5.63 ( <i>s</i> )	117.01	5.62 ( <i>s</i> )	116.90
6	–	141.4	–	141.62	–	142.51	–	142.53
7	5.26 ( <i>s</i> )	72.19	5.27 ( <i>s</i> )	72.60	4.40 ( <i>s</i> )	72.35	4.41 ( <i>s</i> )	73.14
8	2.77 ( <i>dd</i> , $J = 11.9, 2.8$ )	72.39	3.52 ( <i>d</i> , $J = 10.4$ )	77.8	3.58 ( <i>dd</i> , $J = 10.3, 1.6$ )	72.73	3.40 ( <i>dd</i> , $J = 10.6, 2.0$ )	73.13
9	3.69 ( <i>dd</i> , $J = 16.6, 6.8$ )	54.05	3.15 ( <i>br. t.</i> , $J = 10.1$ )	52.77	2.74–2.80 ( <i>m</i> )	52.33	3.13 ( <i>t</i> , $J = 10.3$ )	51.19
10	–	81.4	–	145.5	–	83.01	–	145.98
11	4.78 ( <i>dd</i> , $J = 14.9, 9.8$ )	132.31	5.30 ( <i>dd</i> , $J = 18.1, 13.7$ )	132.35	4.91 ( <i>dd</i> , $J = 15.1, 10.0$ )	132.25	5.21 ( <i>dd</i> , $J = 15.0, 10.3$ )	132.08
12	4.99 ( <i>dd</i> , $J = 14.9, 10.4$ )	129.8	4.70 ( <i>dd</i> , $J = 18.1, 5.5$ )	128.43	4.78 ( <i>dd</i> , $J = 15.0, 9.5$ )	129.36	4.67 ( <i>dd</i> , $J = 15.0, 10.0$ )	128.18
13	3.68 ( <i>dd</i> , $J = 10.4, 1.7$ )	45.8	3.62–3.7 ( <i>m</i> )	45.89	3.58–3.62 ( <i>m</i> )	45.68	3.60–3.68 ( <i>m</i> )	45.72
14	–	207.1	–	206.99	–	206.66	–	206.61
15	–	72.68	–	72.51	–	73.21	–	73.72
16	0.89 ( <i>d</i> , $J = 7.6$ )	16.9	0.89 ( <i>d</i> , $J = 7.2$ )	16.89	0.91 ( <i>d</i> , $J = 7.4$ )	17.09	0.91 ( <i>d</i> , $J = 7.3$ )	17.05
17	1.90 ( <i>s</i> )	17.3	1.93 ( <i>s</i> )	17.44	1.82 ( <i>s</i> )	17.09	1.85 ( <i>s</i> )	17.04
18	1.15 ( <i>s</i> )	29.7	1.68 ( <i>s</i> )	20.65	1.06 ( <i>s</i> )	29.69	1.69 ( <i>s</i> )	29.69
19	0.99 ( <i>s</i> )	24.9	4.69, 4.74 ( <i>2s</i> )	110.79	1.15 ( <i>s</i> )	25.33	4.70, 4.74 ( <i>2s</i> )	110.57
20	1.08 ( <i>d</i> , $J = 7.6$ )	17.3	1.08 ( <i>s</i> )	17.03	1.07 ( <i>d</i> , $J = 7.4$ )	17.08	1.06 ( <i>d</i> , $J = 7.3$ )	20.64
3-AcO	2.10 ( <i>s</i> )	20.7	2.10 ( <i>s</i> )	20.65	2.08 ( <i>s</i> )	20.64	2.07 ( <i>s</i> )	20.80
		170.2		170.09		170.00		170.12
8-MeO	3.48 ( <i>s</i> )	56.55	3.36 ( <i>s</i> )	57.48	3.50 ( <i>s</i> )	56.44	3.38 ( <i>s</i> )	57.71
10-OH	4.58 ( <i>s</i> )	–	–	–	4.50 ( <i>s</i> )	–	–	–
7-AngO	6.15 ( <i>m</i> )	166.3	6.1–6.15 ( <i>m</i> )	166.20	–	–	–	–
	1.95 ( <i>d</i> , $J = 1.4$ )	127.2	1.94 ( <i>d</i> , $J = 1.5$ )	127.45				
	2.03 ( <i>dd</i> , $J = 7.3, 1.4$ )	139.8	2.02 ( <i>dd</i> , $J = 7.2, 1.5$ )	139.48				
		15.9		16.03				
		20.6		20.65				

5-methoxy-3,9,13-trimethyl-10-oxo-15-oxatricyclo[9.3.1.0<sup>1,11</sup>]pentadeca-2,7-dien-4-yl (2*Z*)-2-methylbut-2-enoate.

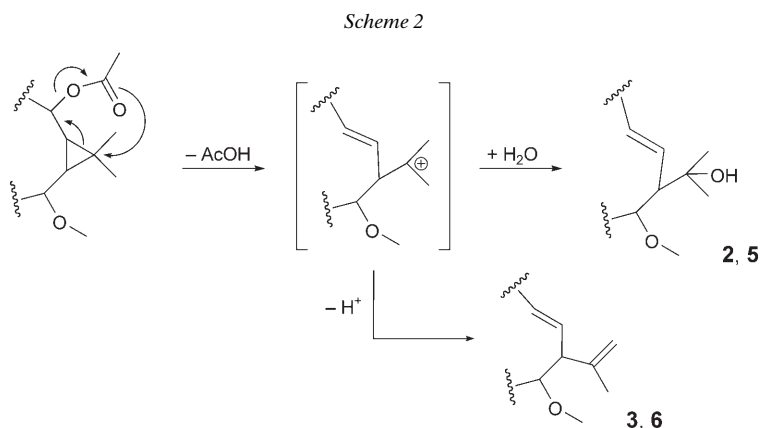
Compound **3** was obtained as a semisolid, with  $[\alpha]_D = -18.0$  ( $c = 0.25$ , CHCl<sub>3</sub>), and analyzed for C<sub>28</sub>H<sub>38</sub>O<sub>7</sub> by HR-FAB-MS [ $m/z$  487.2690 ( $[M + 1]^+$ , calc. 487.2712)]. IR bands at 1739, 1720, and 1705 cm<sup>-1</sup> indicated keto and ester C=O groups. The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra (Table) were similar to those of compound **2**, except for the presence of an isopropylidene group instead of an isopropyl alcohol, as evident from the signals at  $\delta$ (H) 4.69/4.74 (2*s*, CH<sub>2</sub>(19)) and 1.68 (*s*, Me(18)). From these data, the structure of nivulianol B (**3**) was established as (1*S*\*,2*E*,4*R*\*,5*S*\*,7*Z*,9*S*\*,11*R*,13*S*\*,14*S*\*)-14-acetoxy-5-methoxy-3,9,13-trimethyl-6-(1-methylethenyl)-10-oxo-15-oxatricyclo[9.3.1.0<sup>1,11</sup>]pentadeca-2,7-dien-4-yl (2*Z*)-2-methylbut-2-enoate.

Similarly, the structures of compounds **5** and **6** were derived by analyses of their MS and NMR (<sup>1</sup>H, <sup>13</sup>C, COSY, NOESY) data (Table). Nivulianol C (**5**) was obtained as a semisolid,  $[\alpha]_D = +5.83$  ( $c = 0.3$ , CHCl<sub>3</sub>), which analyzed for C<sub>23</sub>H<sub>34</sub>O<sub>7</sub> by HR-FAB-MS

$[m/z$  423.2375 ( $[M + 1]^+$ , calc. 423.2394)], and was identified as (1*R*\*,3*S*\*,4*Z*,7*S*\*,8*R*\*,9*E*,11*S*\*,12*S*\*,13*S*\*)-8-hydroxy-6-(1-hydroxy-1-methylethyl)-7-methoxy-3,9,13-trimethyl-2-oxo-15-oxatricyclo[9.3.1.0<sup>1,11</sup>]pentadeca-4,9-dien-12-yl acetate.

Nivulianol D (**6**), a semisolid with  $[\alpha]_D = -42.85$  ( $c = 0.35$ ,  $\text{CHCl}_3$ ), analyzed for  $\text{C}_{23}\text{H}_{32}\text{O}_6$  by HR-FAB-MS ( $m/z$  405.2286 ( $[M + 1]^+$ , calc. 405.2269)), and was found to correspond to (1*R*\*,3*S*\*,4*Z*,7*S*\*,8*R*\*,9*E*,11*S*\*,12*S*\*,13*S*\*)-8-hydroxy-7-methoxy-3,9,13-trimethyl-6-(1-methylethenyl)-2-oxo-15-oxatricyclo[9.3.1.0<sup>1,11</sup>]pentadeca-4,9-dien-12-yl acetate.

From a mechanistic point of view, the rearrangement of compounds **1** and **4** probably involves an intermediary carbocation formed *via* acetoxy-promoted ring opening followed by loss of AcOH, as proposed in *Scheme 2*.



**2. Biological Activity.** In earlier work [23], we had tested the prostaglandin (PG)  $\text{E}_2$  inhibition properties of several ingol diterpenoids, and found that 12-*O*-acetyl-8-*O*-methyl-7-*O*-(2-methylbut-2-enoyl)ingol exhibits strong *in vitro* anti-inflammatory activity, with an  $\text{IC}_{50}$  value of  $0.003 \mu\text{M}$ , which lies one order of magnitude below that of celecoxib (*Celebrex*<sup>TM</sup>;  $0.050 \mu\text{M}$ ) [23].

The role of cyclooxygenase-2 (COX-2) induction in the pathogenesis of colorectal and breast cancer is well reported [25][26]. Hence, compounds **3–6** were tested for COX-2 inhibition in an *in vitro* enzyme immuno-assay, which involves lipopolysaccharide (LPS)-induced PG  $\text{E}_2$  inhibition. Of all the compounds tested, nivulianol B (**3**) showed 40% inhibitory activity at a concentration of  $10 \mu\text{g/ml}$ . The other compounds did not show significant activity up to  $50 \mu\text{g/ml}$  concentration, nivulianol A (**2**) giving rise to only 10% activity. Hence, nivulianol B (**3**) was chosen for the determination of  $\text{IC}_{50}$ . For a concentration range of  $10–50 \mu\text{g/ml}$ , we determined an  $\text{IC}_{50}$  value of  $36.3 \mu\text{g/ml}$  for **3**.

#### Experimental Part

**Boron Trifluoride Catalyzed Rearrangement.** To a stirred soln. of **1** (25 mg) in anh.  $\text{CH}_2\text{Cl}_2$  (5 ml) under Ar atmosphere,  $\text{BF}_3 \cdot \text{OEt}_2$  (46%;  $80 \mu\text{l}$ ,  $0.56 \text{ mmol}$ ) was added at  $0^\circ$ . The mixture was allowed to warm to r.t. over

1 h. Then, the mixture was diluted with AcOEt, and extracted with sat. aq. NaHCO<sub>3</sub> soln. The org. layer was washed with H<sub>2</sub>O and brine, dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated. The residue was purified by column chromatography (SiO<sub>2</sub>; hexane/AcOEt 8:2) to afford compounds **2** (13.2 mg) and **3** (8.2 mg).

Similarly, with **4** as the starting material, compounds **5** (11.3 mg) and **6** (5.2 mg) were obtained.

*Scandium Triflate Catalyzed Rearrangement.* To a stirred soln. of **1** (25 mg) in anh. CH<sub>2</sub>Cl<sub>2</sub> (5 ml) under Ar atmosphere, Sc·(OTf)<sub>3</sub> (5 mg, 0.01 mmol) was added at r.t., and the mixture was stirred for 1 h. Then, the mixture was diluted with AcOEt, and extracted with sat. aq. NaHCO<sub>3</sub> soln. The org. layer was washed with H<sub>2</sub>O and brine, dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated. Purification by CC (as above) furnished compound **3** (16.5 mg).

Similarly, with **4** as the starting material, compound **6** (13.2 mg) was obtained.

We thank Dr. J. S. Yadav, Director IICT, for his encouragement. V. L. N. R., M. S., and S. M. R are thankful to CSIR, New Delhi, for providing fellowships.

## REFERENCES

- [1] J. E. Backvall, E. V. Bjorkman, L. Petterson, P. Siegbahn, A. Strich, *J. Am. Chem. Soc.* **1985**, *107*, 7408.
- [2] H. M. Walborsky, J. F. Pendleton, *J. Am. Chem. Soc.* **1960**, *82*, 1405.
- [3] A. Delgado, L. Castedo, J. L. Mascarenas, *Org. Lett.* **2002**, *4*, 3091.
- [4] D. H. Gibson, C. H. De Puy, *Chem. Rev.* **1974**, *74*, 605.
- [5] R. H. Crabtree, *Chem. Rev.* **1985**, *85*, 245.
- [6] M. A. Battiste, J. M. Coxon, in 'The Chemistry of the Cyclopropyl Group'; Ed. Z. Rappoport, J. Wiley & Sons, Chichester, 1987.
- [7] P. G. Gassman, S. M. Bonser, *Tetrahedron Lett.* **1983**, *24*, 3431.
- [8] M. R. A. Blomberg, P. E. M. Siegbahn, J. E. Backvall, *J. Am. Chem. Soc.* **1987**, *109*, 4450.
- [9] P. W. Jennings, L. L. Johnson, *Chem. Rev.* **1994**, *94*, 2241.
- [10] P. Kocovsky, J. Srogl, M. Pour, A. Gogoll, *J. Am. Chem. Soc.* **1994**, *116*, 186.
- [11] M. Hayashi, T. Ohmatsu, Y. P. Meng, K. Saigo, *Angew. Chem., Int. Ed.* **1998**, *37*, 837.
- [12] P. A. Wender, C. O. Husfeld, E. Langkopf, J. A. Love, *J. Am. Chem. Soc.* **1998**, *120*, 1940.
- [13] K. Ikura, I. Ryu, N. Kambe, N. Sonoda, *J. Am. Chem. Soc.* **1992**, *114*, 1520.
- [14] I. Ryu, K. Ikura, Y. Tamura, J. Maenaka, A. Ogawa, N. Sonoda, *Synlett* **1994**, 941.
- [15] T. Sugimura, T. Futagawa, A. Mori, I. Ryu, N. Sonoda, A. Tai, *J. Org. Chem.* **1996**, *61*, 6100.
- [16] J. Bayer, R. Madsen, *J. Am. Chem. Soc.* **1998**, *120*, 12137.
- [17] J. O. Hoberg, *J. Org. Chem.* **1997**, *62*, 6615.
- [18] W. R. Lang, C. Djerassi, *Tetrahedron Lett.* **1992**, *23*, 2063.
- [19] V. Ravikanth, P. Ramesh, P. V. Diwan, Y. Venkateswarlu, *Heterocycl. Commun.* **2000**, *6*, 315.
- [20] P. Ramesh, V. L. N. Reddy, N. S. Reddy, Y. Venkateswarlu, *J. Nat. Prod.* **2000**, *63*, 1420.
- [21] N. S. Reddy, T. V. Goud, Y. Venkateswarlu, *J. Chem. Res., Synop.* **2000**, 438.
- [22] V. Ravikanth, V. L. N. Reddy, A. V. Reddy, P. V. Diwan, Y. Venkateswarlu, *Biochem. Syst. Ecol.* **2003**, *31*, 447.
- [23] V. Ravikanth, V. L. N. Reddy, P. Ramesh, T. P. Rao, P. V. Diwan, Y. Venkateswarlu, *Phytochemistry* **2002**, *59*, 331.
- [24] V. Ravikanth, V. L. N. Reddy, A. V. Reddy, K. Ravinder, T. P. Rao, T. S. Ram, K. A. Kumar, P. V. Diwan, Y. Venkateswarlu, *Chem. Pharm. Bull.* **2003**, *51*, 431.
- [25] P. Paola, *Toxicol. Lett.* **2000**, *112–113*, 493.
- [26] G. Singh-Ranger, K. Mokbel, *Eur. J. Surg. Oncol.* **2002**, *28*, 729.

Received April 12, 2005